

## **Inorganic Compounds**

- Lack carbon
- tend to be small, simple molecules
- May be ionic or covalent
- examples
  - water
  - salts
  - many acids and bases

#### Water

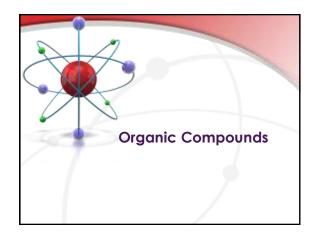
- Polar molecule
- High heat capacity
- "Universal" solvent
- Chemical reactivity
- Cushioning

### Salts

- ionic compounds
- The salts of many metallic elements are found throughout the body such as those containing:
  - Calcium
  - Phosphorus

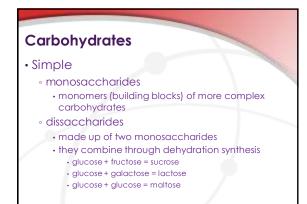
### Salts

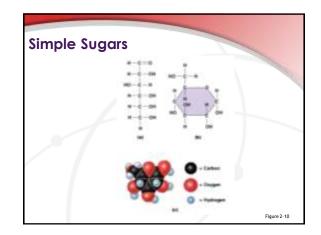
- Dissociate in water
  - Sodium and potassium ions are important for nerve impulses
  - iron forms part of hemoglobin which transports oxygen in the blood
- electrolytes
  - substances that conduct electrical current in solution

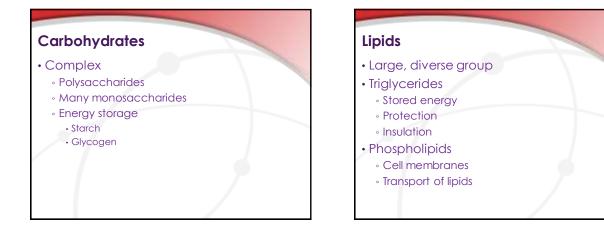


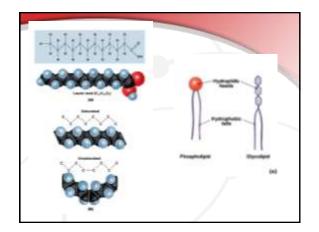
## Carbohydrates

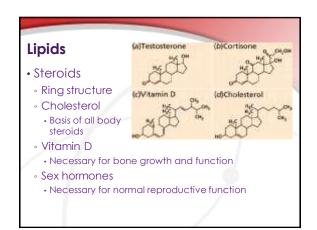
- primary source of chemical energy
- sugars and starches





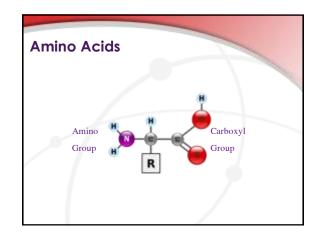






#### **Proteins**

- Over 50% of organic matter in body
- High variable function
  - Structure
  - Transport
  - Antibodies
  - Enzymes
  - Contraction
  - Hormones
  - Storage
- Combinations of amino acids



### Enzymes

- Biological catalysts
  - substance that increases the rate of a reaction without being used up in the process

# **Nucleic Acids**

- transmit hereditary materials from one generation to another
- directors of the cell's activities

